2007 Vol. 9, No. 22 4403-4406

3,3'-Dipyridyl BINOL Ligands. Synthesis and Application in Enantioselective Addition of Et₂Zn to Aldehydes[#]

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Received May 30, 2007

ABSTRACT

O Et₂Zn OH OH OH OH OH OH
$$(R)$$
-4 (R) -6 (R) -7 (R) -8 (R) -9 (R) -9

3,3'-Di(2-pyridyl) BINOLs (4), prepared in three steps from commercial substances by a combined directed ortho metalation—Negishi cross-coupling protocol, constitute new members of the BINOL ligand family. They are among the most selective, general ligands, providing equal or greater stereoselectivities in the Et₂Zn addition to benzaldehydes and cinnamaldehydes (Table 3) than any single BINOL ligand reported to date.

We report on the convenient synthesis of new 3,3'-di(2-pyridyl) BINOL ligands 4a-h by the directed *ortho* metalation (DoM)—cross-coupling protocol developed in our laboratories¹ and demonstrate their application to the highly enantioselective addition of Et_2Zn to benzaldehydes and cinnamaldehydes. Since the pioneering studies of Cram,² the

BINOL motif has enjoyed widespread use as a chiral element/ligand scaffold in asymmetric synthesis.³ Although structural variations of the BINOL 3,3'-substituents have mainly been concerned with steric effects,⁴ an increasing number of reports deal with systems capable of metal coordination for application in a range of reactions⁵ among which, most prominently, are the diorganozinc additions to aldehydes.⁶ Our new family of dipyridyl BINOL ligands **4a—h** show enantioinductions in this reaction which are comparable to or higher than those previously observed for the 3,3'-diamide BINOL **1** by Katsuki^{6b,h,i} and the 3,3'-bisaryl BINOL **2** by Pu^{6a,e—g} (Scheme 1).

Of the two categories of 3,3'-substituted BINOLs which are capable of coordination (Scheme 2), simple Lewis acid $(\mathbf{A})^7$ or bifunctional (\mathbf{B}) ,8 we propose that the dipyridyl BINOLs act as bifunctional catalysts9 and that the reported

^{*}Dedicated to Hisashi Yamamoto for lasting contributions to our armamentarium of synthetic methods, particularly in Lewis acid catalysis, for thoughtful service to high standards of our literature, and for warm friendship.

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Scheme 1

CONR₂
OH
OH
CONR₂
1:
$$R = Me$$
, Et, Bu
RO
OH
OH
OH
2: $R = C_6H_{13}$

results convey knowledge for the design and development of new robust BINOL ligands for asymmetric catalysis.

The combined DoM—cross-coupling sequence to chiral 3,3'-disubstituted BINOLs using exchangeable 3,3'-dihalo and dimetalo 2,2'-di-DMG-bearing partners (DMG = directed metalation group: OMOM, OCONEt₂, OSEM) is a particularly flexible, rapid, and convergent route to optically

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A: Lewis Acids



B: Bifunctional Catalysts

pure ligands.^{10,11} After experimentation with combinations of cross coupling partners,¹² the Negishi regimen involving reaction of the 3,3'-di-ZnCl BINOLs with 2-halopyridines under Pd(PPh₃)₄ catalysis provided optimal conditions to afford coupled products which, for efficiency and ease of handling, were directly deprotected to give the 3,3'-di(2-pyridyl) BINOLs 4 in acceptable yields along with modest yields of the monocoupled products (Table 1). Use of

Table 1. Synthesis of 3,3'-Di(2-pyridyl) BINOL Ligands 4a-h

entry	X, G	ligand	yield, %	entry	X, G	ligand	yield, %
1 2 3 4	Br, H I, 3-OC ₄ H ₉ I, 3-OC ₆ H ₁₃ I, 3-OCH ₂ Ph	4a 4b 4c 4d	75 45 42 49	5 6 7 8	Br, Me I, 3-F I, 4-t-Bu 2-Br, quinoline	4e 4f 4g 4h	41 60 68 66

 π -acidic ligands (P(furyl)₃) did not result in improved yields. Although the intermediate 3,3'-di-ZnCl BINOL may be prepared directly through DoM to form 3,3'-di-Li BINOL then transmetalation with ZnCl₂ and the 3,3'-di-ZnCl BINOL

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thus generated readily participates in Negishi cross coupling, significant monocoupled products result due to incomplete BINOL lithiation. We therefore found it advantageous to apply a two-step sequence, *via* intermediate iodide 3, which allows for rejection of the monolithiated products formed during the DoM step, facilitates clean formation of 3,3'-di-ZnCl BINOL, and significantly reduces the amount of monocoupled product.

For our studies, we employed the enantioselective addition of Et₂Zn to aldehydes, a thoroughly explored reaction, ¹³ which has become recognized as a standard assay in guiding new ligand development. ¹⁴ The results (Table 2) demonstrate

Table 2. Asymmetric Et_2Zn Addition to Benzaldehyde and Cinnamaldehyde Catalyzed by 3,3'-Di(2-pyridyl) BINOL Ligands 4a-h

$$\begin{array}{c} O \\ R \end{array} \begin{array}{c} Et_2Zn \\ 5 \hspace{0.1cm} mol\% \hspace{0.1cm} (R)\text{--4} \\ 2h \end{array} \begin{array}{c} OH \\ R \end{array} \begin{array}{c} OH \\ OH \\ OH \end{array}$$

Entry	Ligand	G	R = Ph Yld, $%^a$ % ee ^b		R = % Ph	
					Yld, % ^a % ee ^b	
1	4a	Н	91	98	77	93
2	4b	3-OC ₄ H ₉	71	99	78	92
3	4c	3-OC ₆ H ₁₃	96	98	92	96
4	4d	3-OCH ₂ Ph	81	97	83	90
5	4e	3-Me	63	97	74	89
6	4f	3-F	56	97	52	92
7	4g	4- <i>t</i> -Bu	86	96	85	89
8	4h	2-quinoline	56	94	72	88

^a Determined by GC analysis using undecane as an external standard. ^b Determined via chiral GC analysis using β-cyclodextrin stationary phase. (See the Supporting Information.)

that the series of 3,3'-di(2-pyridyl) BINOLs **4a**—**h** are very effective ligands for the catalytic asymmetric addition of Et₂-Zn to benzaldehyde (the reaction standard) and cinnamaldehyde (the reaction benchmark, see below). Thus, compared to the unsubstituted pyridyl ligand **4a** (entry 1), 3-oxygenated pyridyl derivatives **4b**—**d** impart similar (entries 2 and 4) or improved (entry 3) enantioselectivities, while 4-substitution (entry 7) and aromatic ring annelation (2-

quinolinyl, **4h**, entry 8) result in similar or decreased enantioselectivities.

Substrate scope was then examined using the most promising ligands: **4a** and **4c** (Table 3). Ligand **4a** showed

Table 3. Asymmetric Et_2Zn Addition to Aldehydes Catalyzed by 3,3'-Di(2-pyridyl) BINOL Ligands **4a** and **4c**

Entry			d (<i>R</i>) -4a ^a % ee ^b	Ligands Yld, % ^a	
1	PhCHO	91	98 (<i>R</i>) ^c	96	98 (R)°
2	Ph CHO	77	93 (R) ^c	92	96 (R) ^c
3 4 Me0	CHO 2-OMe 4-OMe	97 31	95 97	99 4 0	97 98
5	CHO	37	96	25	97

^a Determined by GC analysis using undecane as an external standard. ^b Determined by chiral GC analysis using β -cyclodextrin stationary phase. ^c Assessed by comparison of the measured optical rotation with known values. (See the Supporting Information.)

excellent enantioinduction for a range of aldehydes albeit, in some cases (entries 4 and 5), with decreased yields while ligand 4c displayed significantly improved selectivity (>97% ee) for both *ortho*-substituted aryl (entry 3) and α,β unsaturated (entry 2) aldehydes. In comparison with the Katsuki diamide 16b,h,i and Pu bis aryl 26a,e-g ligands, the following features of the dipyridyl ligands 4a and 4c are noted: (1) Ligand 4a displays stereoselectivities of similar magnitude as 1 and 2, while the selectivity of ligand 4c is consistently higher; (2) catalytic activities of 4a and 4c appear to be lower than that of 2 since the aliphatic aldehydes did not undergo ethylation and less reactive aldehydes (entries 4 and 5) required extended reaction times for complete conversion; (3) prominent solvent effects were noted (THF > PhMe > CH₂Cl₂), which are consistent with those observed for ligands $\mathbf{1}^{6b,h,i}$ and $\mathbf{2}^{6f}$ but opposite to those of

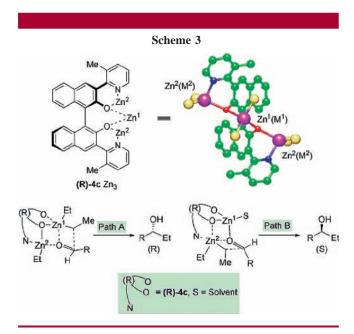
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⁽¹¹⁾ Cram first demonstrated the DoM—cross-coupling route for the synthesis of 3,3'-substituted BINOLs from 2,2'-diOMe BINOLs; see ref 2. The limitations placed on this route by the harsh conditions required for OMe deprotection to the corresponding BINOLs have been alleviated by the application of other DMGs which may be cleaved under mild conditions, see ref 1.

⁽¹²⁾ Based on our experience, cross coupling of heteroaryls is best achieved using the heteroaryl halide and the corresponding Suzuki, Stille, or Negishi organometallic partners which takes advantage of the stability and availability of the former and avoids the problematic formation and instability of the latter.

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the amino alcohol class of ligands; 15 and (4) the R ligand consistently gives the R alcohol; hence, the sense of stereoinduction is consistent with the other 3,3'-disubstituted BINOL ligands.

For the purpose of mechanistic rationalization, two models are considered (Scheme 2): simple Lewis acid type $(A)^7$ in which the 3,3'-substituents coordinate the central metal of the complexes, imparting further rigidity and propagating the stereochemical environment around the metal, and a complex capable of bifunctional catalysis (**B**)⁸ by formation of aggregates or coordination of the reactant to direct it toward the substrate bound to the Lewis acidic central metal. The matching stereoselectivity profiles, direction of enantioinduction, generality, and solvent effects displayed by 1,6b,h,i 2,6f and 4a-h suggest the participation of all ligands via a common mechanism. There is, however, a limited amount of physical and theoretical data accumulated on these metal-ligand complexes and the mechanisms thus far proposed, involving dimeric complexes (A) or bifunctional catalysis (B), are largely speculative. In the absence of such data, we are forced to rely largely on analogy with the β -amino alcohol catalyzed reaction¹⁶ to formulate a working hypothesis for the mode in which these ligands function in enantioselective catalysis. The studies (¹H NMR and X-ray structure) of the zinc complex of 1 disclosed by Katsuki^{6b,h,i} suggest that all 3,3'-disubstituted BINOL ligands 1, 2, and

4a−h catalyze the Et₂Zn addition to aldehydes *via* complexes of type (**B**) (Scheme 2). Although assignment of M¹ or M² to Lewis acidic or nucleophilic roles in the catalyst complex is premature at present, application of a geometry-optimized model (generated via PM3 calculations) of (R)-4c Zn₃ complex (Scheme 3) and invoking the accepted transition state for the Et₂Zn addition to aldehydes¹⁶ suggests the assignment of M² to a Lewis acidic role and M¹ to the nucleophilic role in the reaction, and this is in accord with the observed direction of stereoinduction (path A). The alternative configuration (path B), in which the aldehyde is bound to the central M¹ and the ethyl is delivered by M², leads to the incorrect prediction of stereoinduction. Based on this argument, we further surmise that the improved stereoselectivity observed using 3,3'-dipyridyl BINOLs is the result of angle alternation of the pyridine-biaryl axis which is dependent on the 3,3'-substituents with the benzyloxy and methyl substitution being too large while the alkoxy substituent being optimal to allow geometrically proper coordination.

In summary, the unsubstituted 3,3'-dipyridyl BINOL ligand 4a, readily prepared in two steps from commercially available starting materials, delivers ee's in Et₂Zn addition to aryl and cinnamyl aldehydes equal to those observed for the more complex ligands 1 and 2. The more elaborate ligand **4c** delivers ee's superior to any of the BINOL-type ligands thus far reported, and both 4a and 4c maintain the unique generality displayed by the family of BINOL ligands for Et₂-Zn addition to both aromatic and α,β -unsaturated aldehydes. A model is proposed which is in agreement with the sense of the observed enantioinduction. Our mechanistic understanding of the enantiodiscrimination in the Et₂Zn addition to aldehydes catalyzed by 3,3'-disubstituted BINOL ligands may be further enhanced from the investigation of a wider range of 3,3'-disubstituted ligands incorporating donor substituents which promote bifunctional catalytic behavior in conjunction with computational studies.

Acknowledgment. We thank NSERC Canada for support. R.R.M. is grateful to Shire Pharmaceuticals (formerly Biochem Pharma) for an NSERC/IPS Fellowship, and O.P. thanks the DFG (Deutsche Forschungsgemeinschaft) for a fellowship (Grant No. Pr526/1-1). We thank HEC for a foreign scholarship to S.M.S.H. and are grateful to Tom Blackburn for his moral support.

Supporting Information Available: Detailed experimental procedures and spectroscopic data for compounds 3 and 4a-h. This material is available free of charge via the Internet at http://pubs.acs.org.

OL071276F

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